Ultra-High Temperature Ceramic Composites for Leading Edges - 2004

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Ultrahigh temperature ceramics (UHTC) have performed unreliably due to material flaws and attachment design. These deficiencies are brought to the fore by the low fracture toughness and thermal shock resistance of UHTC. If these deficiencies are overcome, we are still faced with poor oxidation resistance as a limitation on UHTC applicability to reusable launch vehicles. We have been addressing the deficiencies of UHTC for the past two years via a small task at GRC that is in the Airframe part of the Next Generation Launch Technology Program. Our focus is on composite constructions and functional grading to address the mechanical issues and on composition modification to address the oxidation issue. The progress on approaches to improving oxidation resistance by alloying and functional grading will be reported.

Keywords: ceramics, composites, carbon fibers, borides, oxidation

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Ultra-High Temperature Ceramic Composites for Leading Edges

OUTLINE

- Background
- Objective
- Ta Modification of ZrB_2 - SiC
  - Furnace oxidation
  - Arc jet test
- UHTCC
  - Processing
  - Characterization of Starfire UHTCC
- Concluding Remarks

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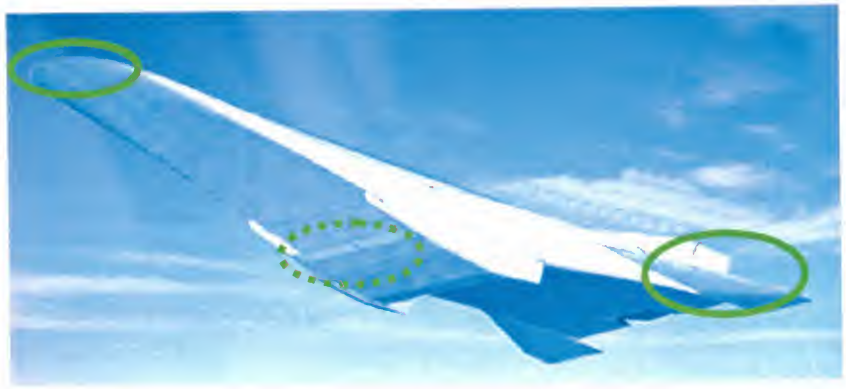
Ultra High Temperature Ceramic Composites (UHTCC) for Leading Edges

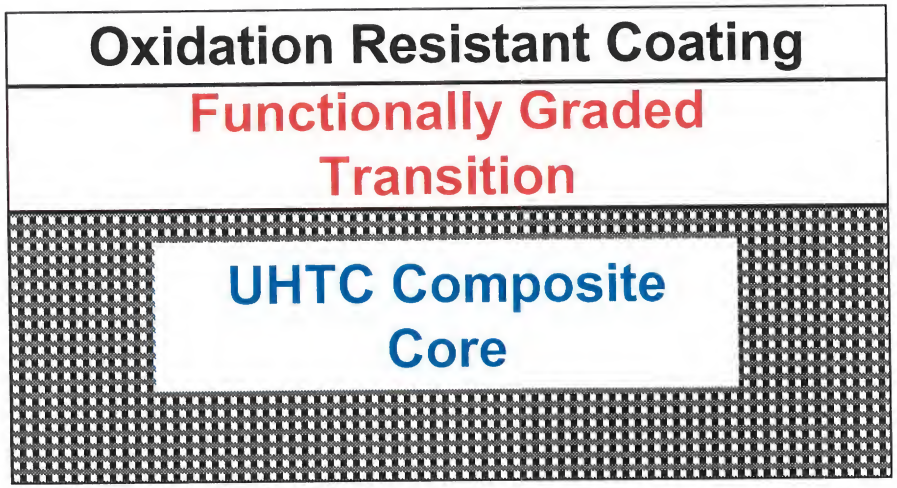


Key Issues

- Thermal stress resistance
- Oxidation resistance
- Temperature capability
- Architecture optimization

Objective

- Develop UHTCC for ≥3600°F, reusable, hypersonic vehicle leading edges



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Mechanisms for ZrO_2 Protective Scale Enhancement via Ta_2O_5 Additions

- **Ta_2O_5 as a glass modifier**
  - Talmy et al
- **Ta^{+5} as a dopant in ZrO_2 lattice**
  - Fill O^{-2} vacancies
  - Decrease O^{-2} transport
- **Ta_2O_5 as a major oxide scale constituent**
  - Phase V
  - Low melting point

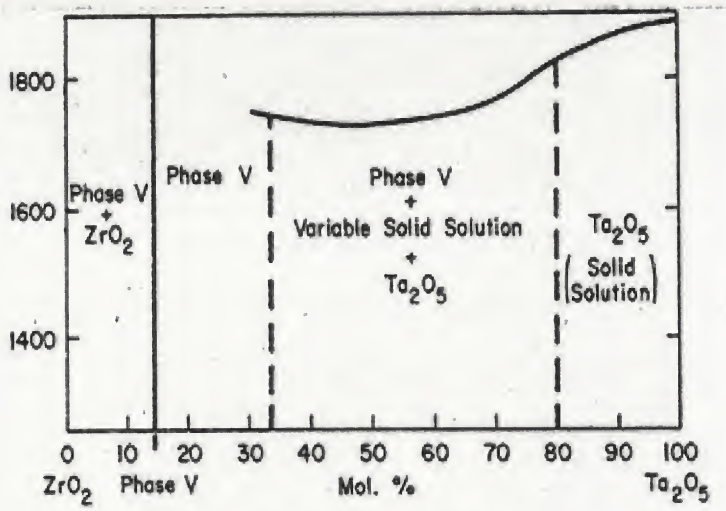

FIG. 374.—System Ta_2O_5–ZrO_2.

B. W. King, John Schultz, E. A. Durbin, and W. H. Duckworth, U. S. Atomic Energy Comm., BMI-1106, 15 (1956).

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Materials Studied

COMPOSITION	DESIGNATION
ZrB_2 + 20v/o SiC	ZS
HfB_2 + 20v/o SiC	HS
ZrB_2 + 20v/o SiC + 20v/o $TaSi_2$	ZSTS
ZrB_2 + 33v/o SiC	ZS+
ZrB_2 + 20v/o SiC + 20v/o TaC	ZSTC
HfB_2 + 20v/o SiC + 20v/o $TaSi_2$	HSTS

- powders ball milled, hot pressed ~ 2000°C, 10 ksi, 2h, vacuum
- 2.54 x 1.2 x 0.3 cm coupons machined from hot pressed plates
- coupons cleaned, weighed, surface area determined

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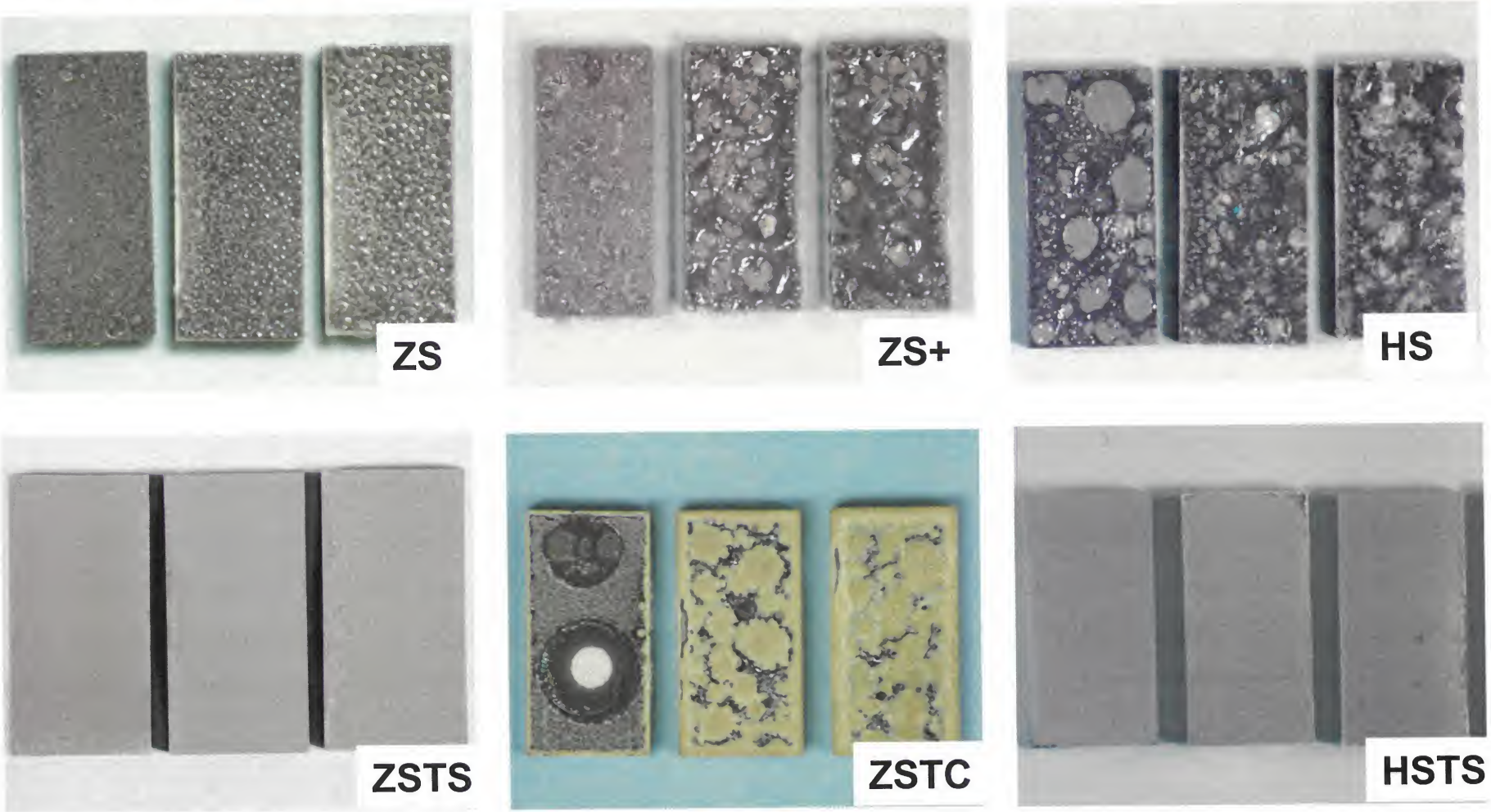
UHTC Coupon Appearance After Oxidation
1627°C, Stagnant Air



Left to right: 1, 5, and 10 ten-minute cycles

Ta-containing compositions show less glass formation

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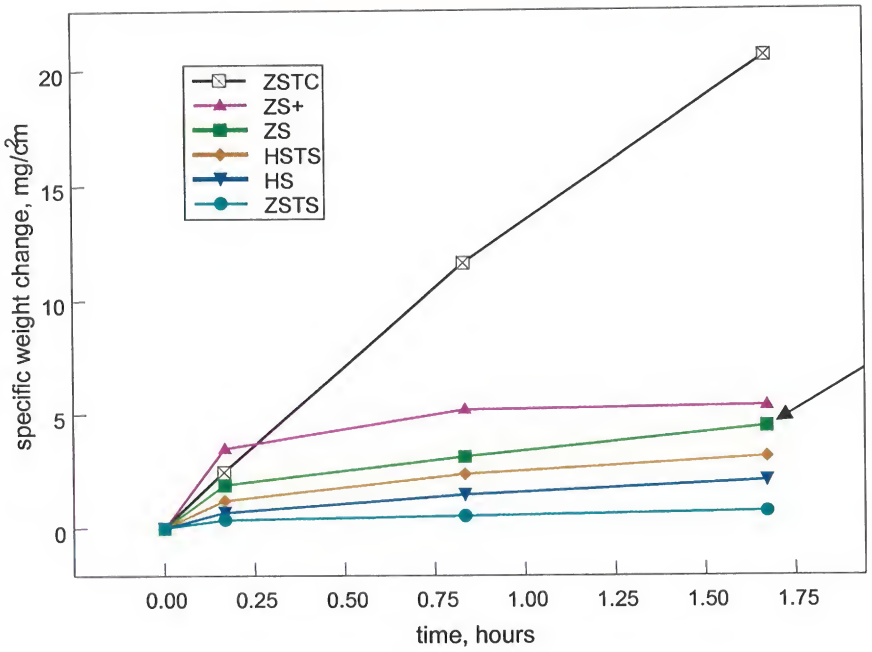
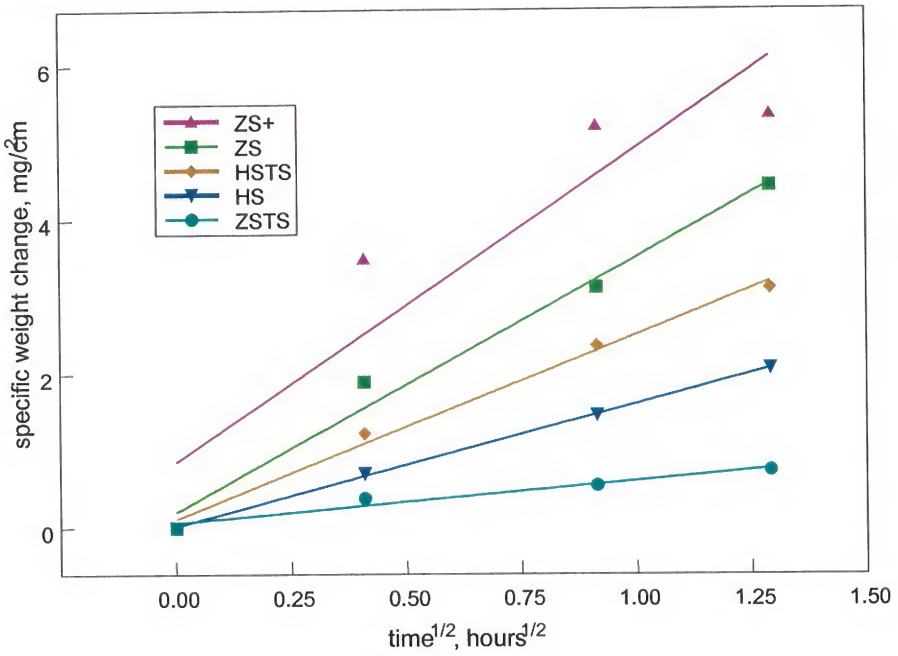
Weight Change of UHTC After Oxidation 1627°C, Stagnant Air





- Poor results for ZS+ relative to ZSTS show that Ta-addition, not additional Si, is responsible for improved oxidation resistance of ZSTS.
- TaC additions do not improve oxidation resistance at 1627°C.
- $TaSi_2$ additions do not improve HS.

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XRD Results for UHTC After Oxidation at 1627°C in Stagnant Air for 100 Min.

composition	phases present
ZS	ZrO_2(m), **ZrO_2(c)**
HS	HfO_2(m), **HfO_2(c)**, **$HfSiO_4$**
ZSTS	ZrO_2(m), **ZrO_2(c)**
ZS+	ZrO_2(m), **ZrO_2(c)**
ZSTC*	ZrO_2(t), **ZrO_2(m)**
HSTS	HfO_2(m), HfO_2(c), $HfSiO_4$

bold = major phase

$Ta_2O_5 \cdot 6ZrO_2$ (orthorhombic) difficult to distinguish from ZrO_2(c)

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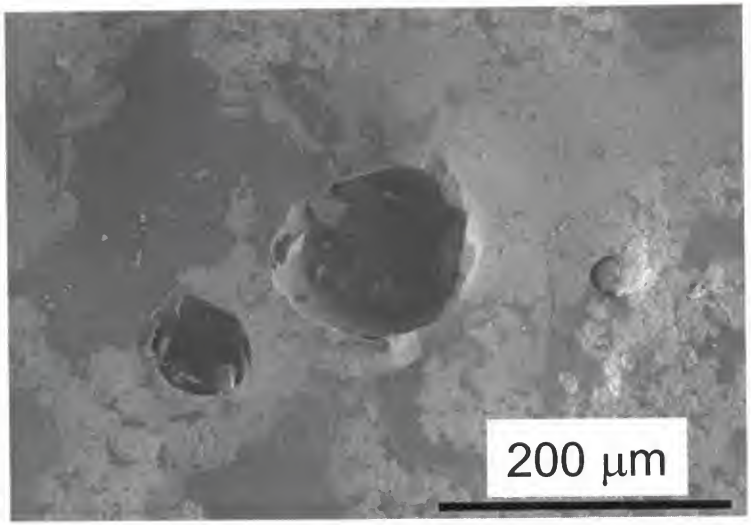
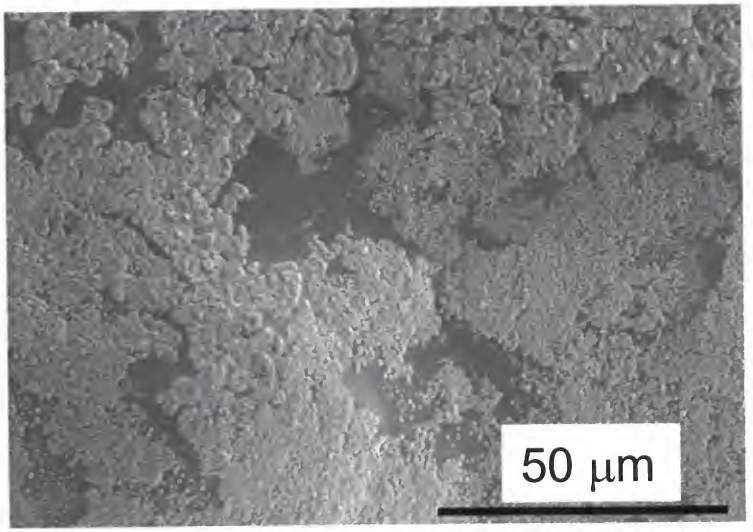
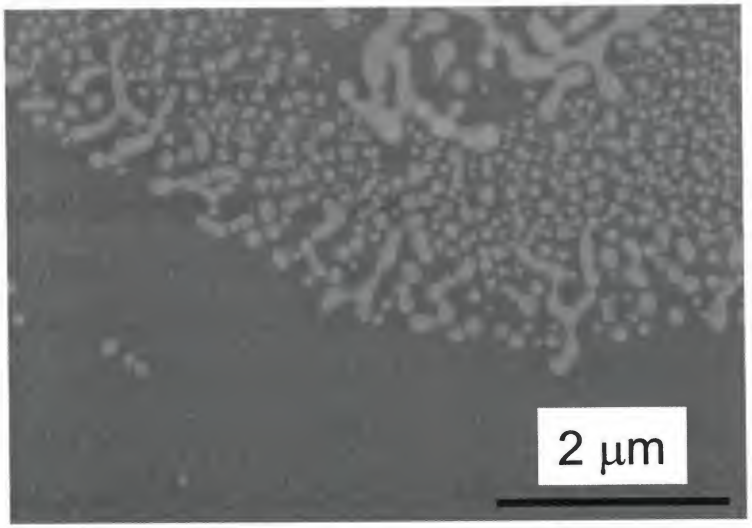
Surface Microstructure of ZS After Oxidation at 1627°C in Stagnant Air







- Non-uniform dispersion of surface oxide phases.
- Fine dispersion of second phase in glassy areas.

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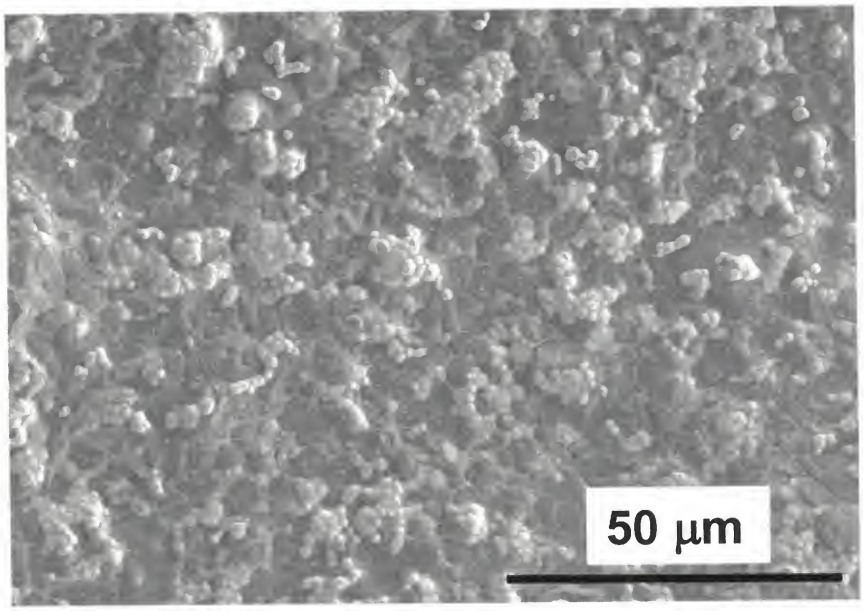
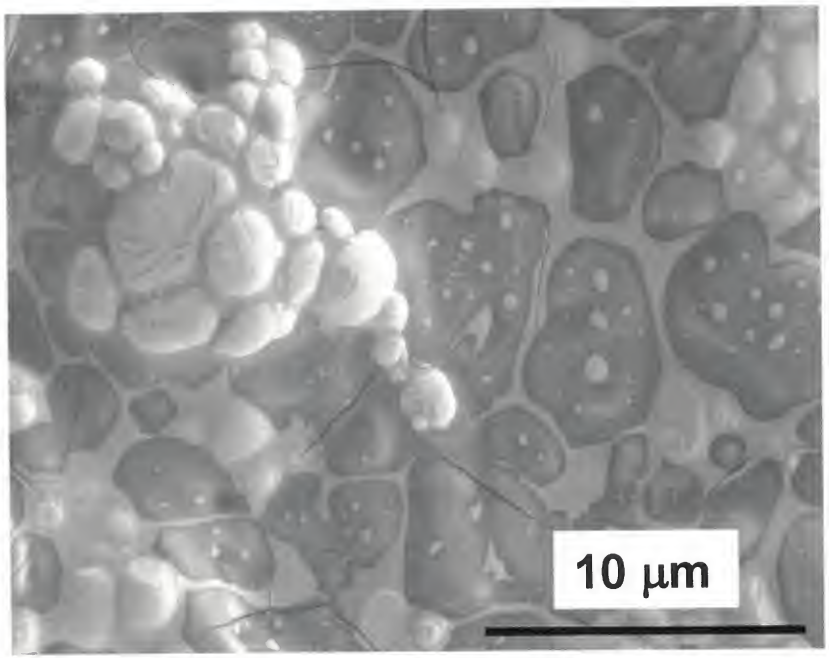
Surface Microstructure of ZSTS After Oxidation at 1627°C in Stagnant Air





- Uniform dispersion of surface oxides
- Evidence of glass immiscibility

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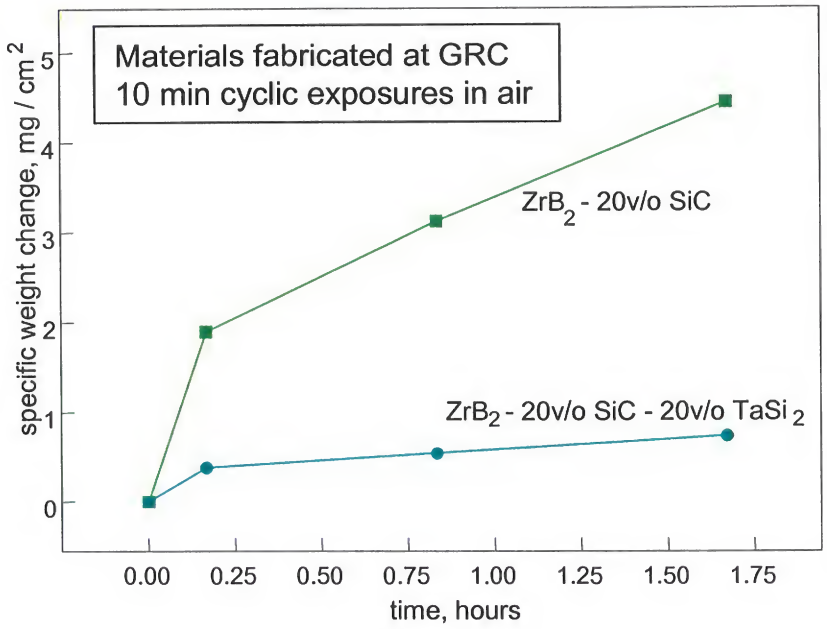
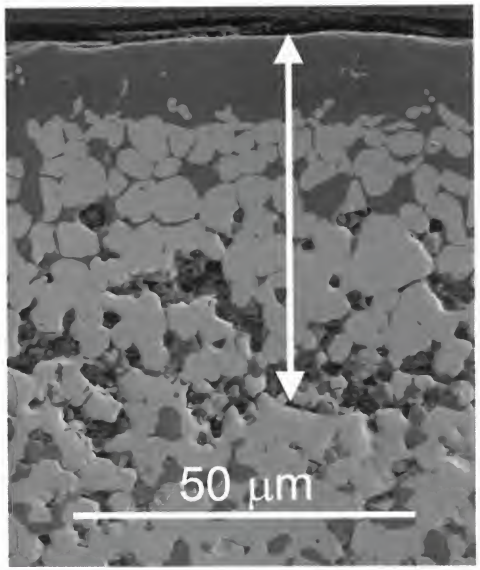
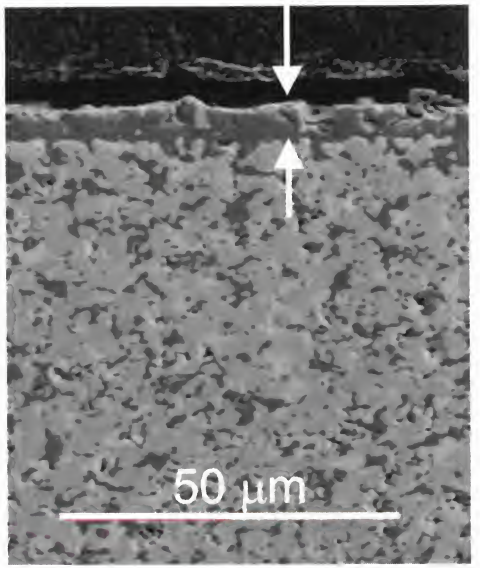
Tantalum Additions Improve Oxidation Resistance of ZrB_2-20v/o SiC at 1627°C





ZrB_2 – 20v/o SiC



ZrB_2 – 20v/o SiC – 20v/o $TaSi_2$

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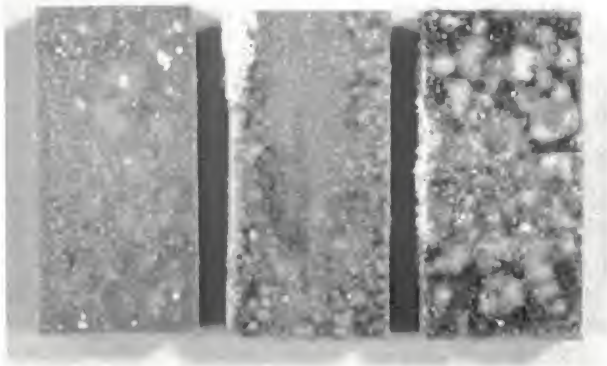
UHTC Furnace Oxidized in Air

	ZrB_2 + 20 v/o SiC			ZrB_2 + 20 v/o SiC + 20 v/o $TaSi_2$		
	1*	5	10	1	5	10
1627°C						
1927°C						

5 cycle specimen after attempted removal from setter

* # of cycles (1 cycle = 10 minutes hot & 10 minutes cool)

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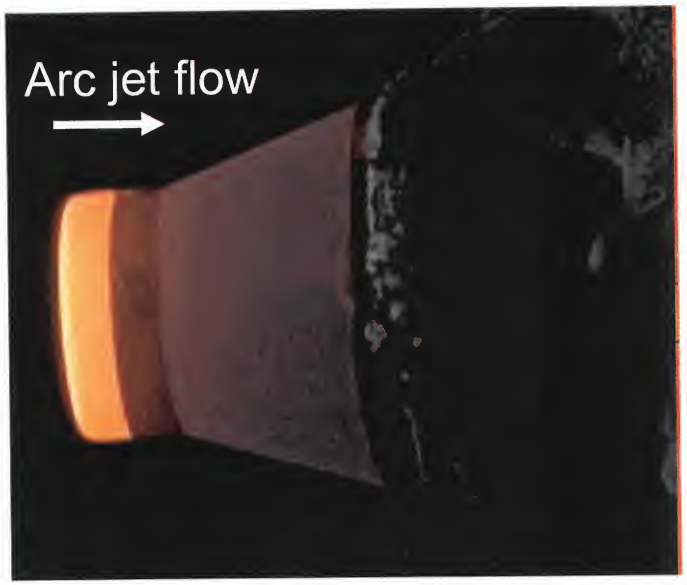
Arc Jet Test of ZrB_2 - 20 v/o SiC - 20 v/o $TaSi_2$

2.54 cm
Specimen

2.54 cm
Specimen mounted in graphite holder



Specimen and graphite holder mounted in arc jet sting

Test Conditions

350 W/cm²
0.07 atm.
600 seconds
% Δwt = -1.4

Temperatures and Heat Flux

Measured Temperature ~1800 °C
Edge Heat Flux Estimate ~ 600 W/cm²
Edge Temp Estimate ~1950 to 2000°C

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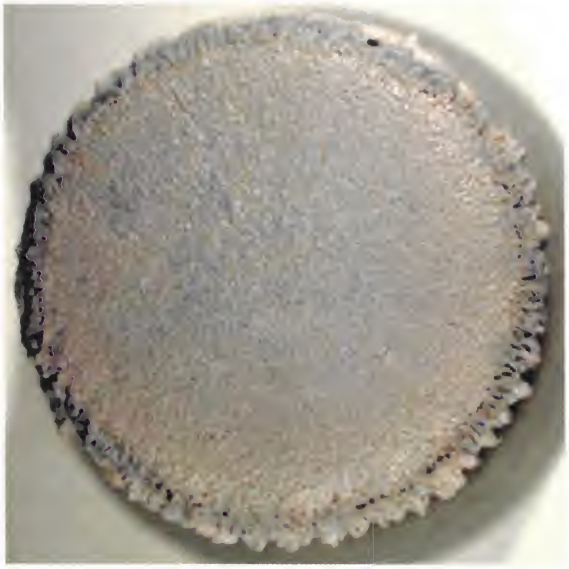
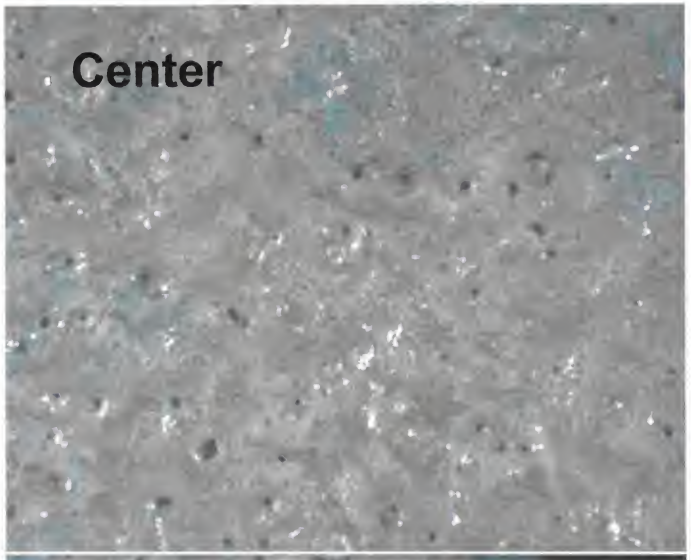
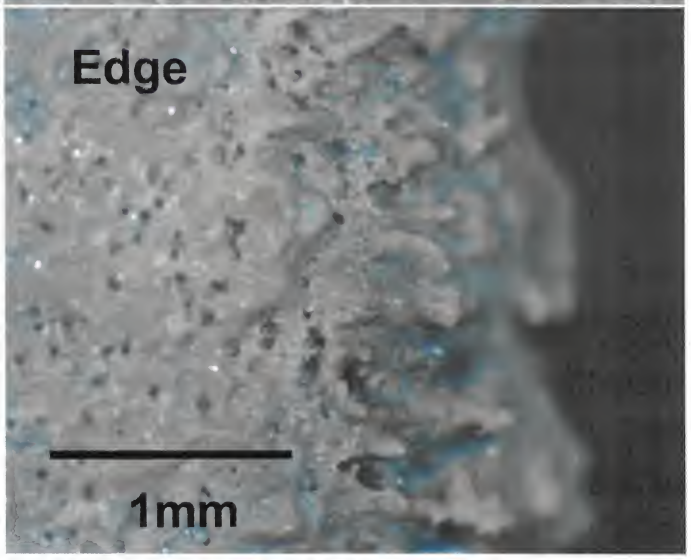
Images of GRC Model After Arc Jet Test



XRD Results
- ZrO_2 (m)
- ZrO_2 (c)
- Ta_2O_5





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Some Answers

- Is Ta or Si in $TaSi_2$ responsible for the improved oxidation resistance of ZrB_2 – 20v/o SiC – 20v/o $TaSi_2$? Tantalum.
- Does the benefit of $TaSi_2$ additions observed at 1627°C extend to higher temperatures? No, not at 20v/o addition levels.
- Are $TaSi_2$ additions the best way to add Ta to ZrB_2-based UHTC? TaC additions are not effective at 1627°C.
- Can $TaSi_2$ additions improve the oxidation resistance of HfB_2-based UHTC which are already superior to ZrB_2-based materials? No, not at 20v/o addition levels.
- Can anything be learned about the mechanism by which Ta-additions improve the oxidation resistance of ZrB_2-SiC materials at 1627°C? Tantalum effects both glass phase and ZrO_2.

OBJECTIVES

- Characterize a UHTC composite plate fabricated by Starfire Systems, Inc.
  - CAVEAT: Recognize that little or no development effort went into fabrication of this material. It was a best effort fabrication for NASA LaRC
- Reveal some of the issues associated with the UHTCC concept

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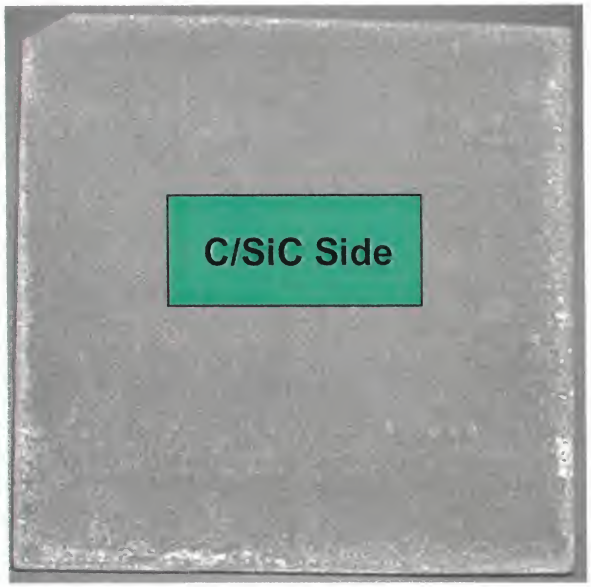
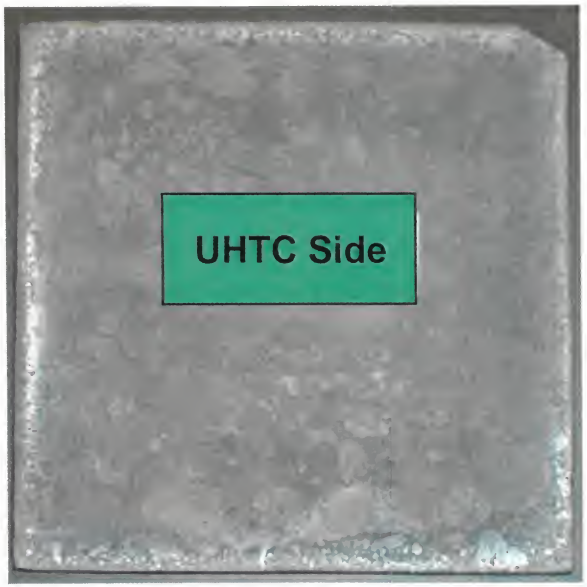
Starfire UHTCC Plate

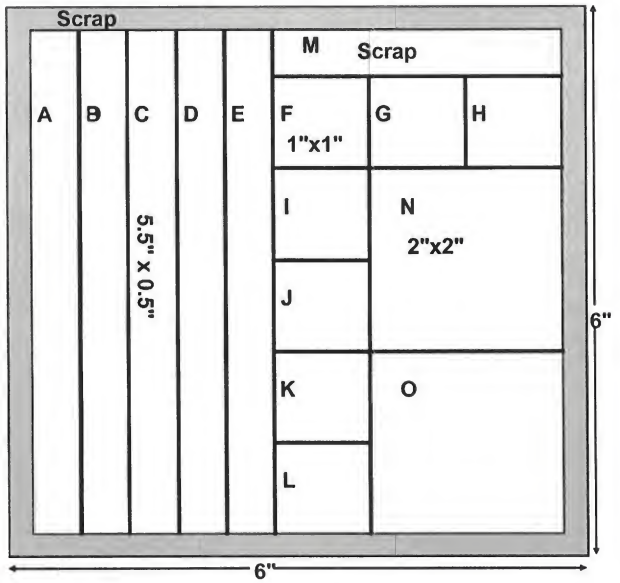
- **Constituents**
  - **Zoltek Panex® 30 Carbon Fabric PW06**
  - **Starfire Systems' SP-Matrix Polymer (Allylhydridopolycarbosilane (AHPCS))**
  - **HfB_2 Powder**
  - **SiC Powder**

Processing of Part Number 000928-6-64

- **Initial Cycle:**
  - For the initial lay up the **bottom 6 layers of cloth are coated with a SiC/AHPCS** slurry and the **top 5 layers are coated with a HfB_2 /AHPCS slurry** and assembled in a mold.
  - Cured to 400°C and fired to 850°C under inert gas to pyrolyze.
- **Cycle 2:**
  - **Coat the HfB_2 side of the plate with more HfB_2 /AHPCS slurry.**
  - Fire directly to 850°C under inert gas.
- **Cycle 3: Repeat cycle 2**
- **Cycle 4 - 10:**
  - **Vacuum infiltrate with AHPCS only.**
  - Pyrolyze directly to 850°C under inert gas, no clamping necessary.







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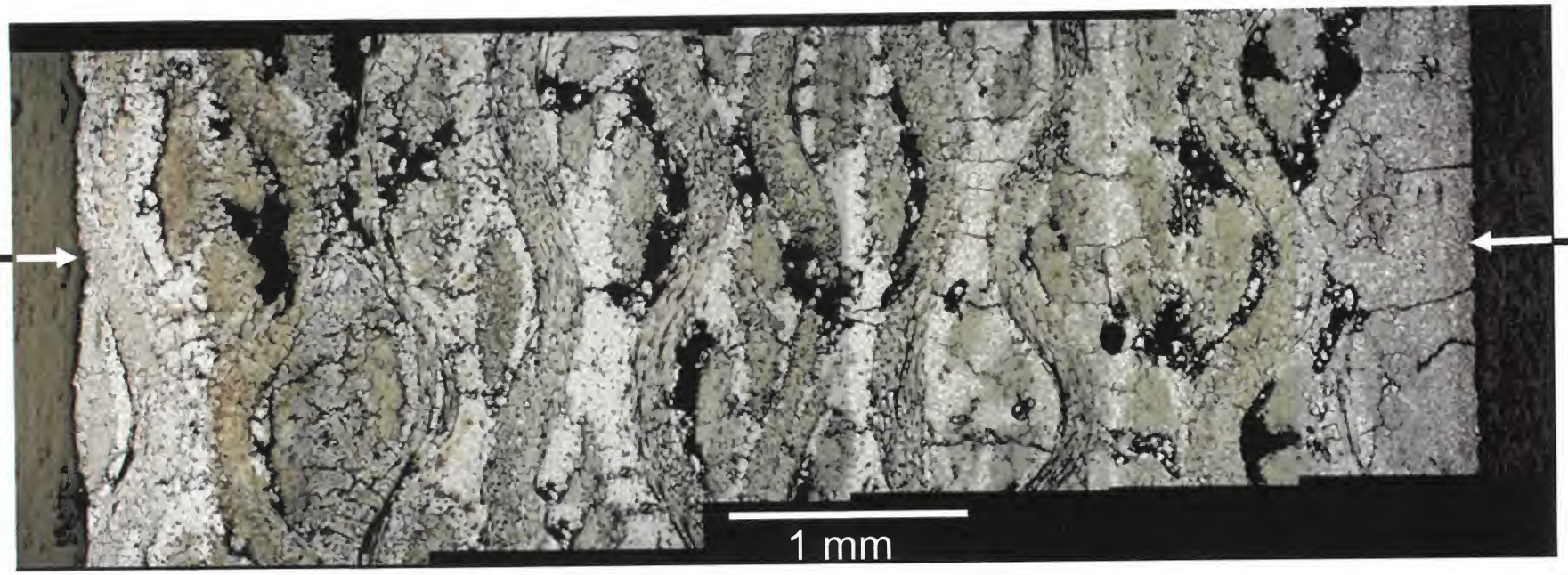
UHTCC Cross-section



SiC
C

Phases Detected by XRD

HfB_2

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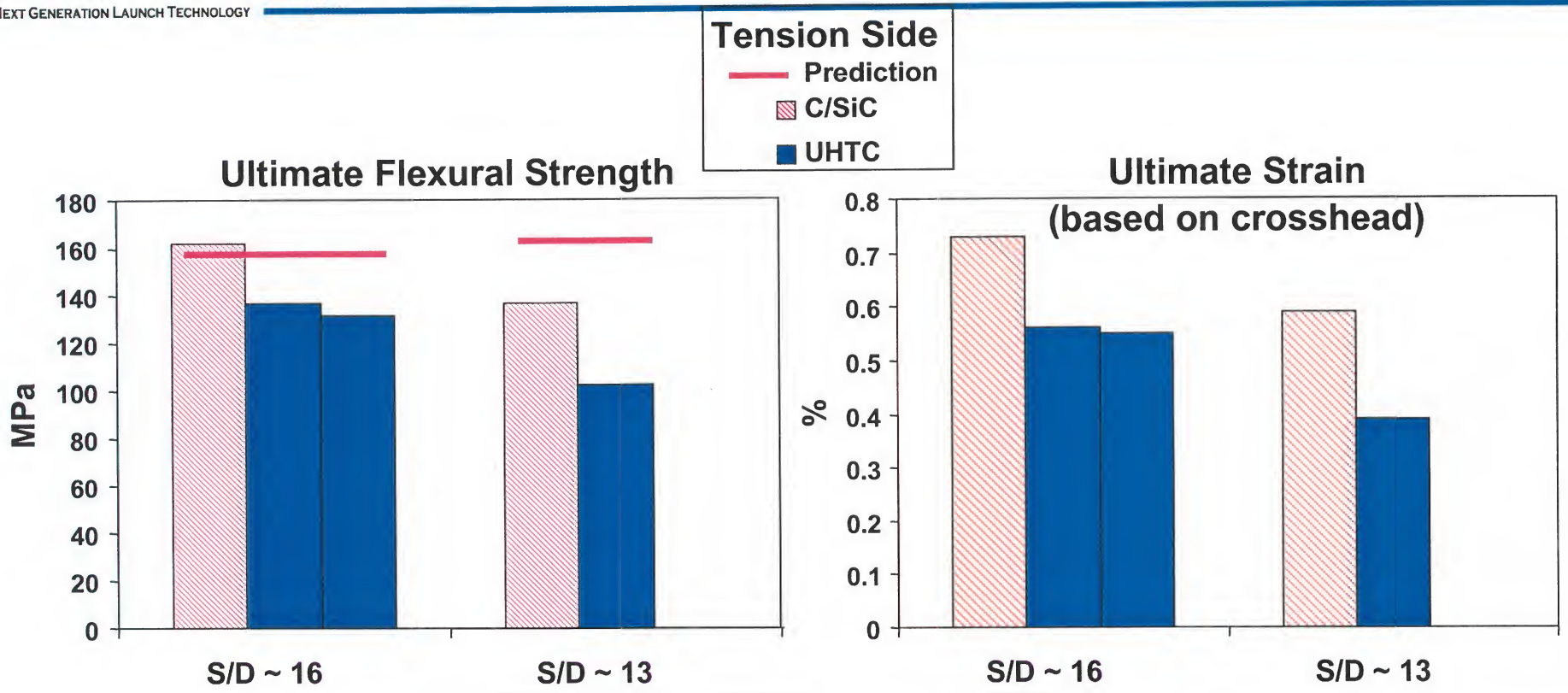
UHTCC Flexural Strength Results



Note: S/D ≥ 16 necessary for valid 4 point flexural test

Predictions Based on Beam Theory

- Calculated load at 0.7% strain
- Panex 30 minimum property: E = 193 GPa
- Rule of mixtures with no matrix contribution

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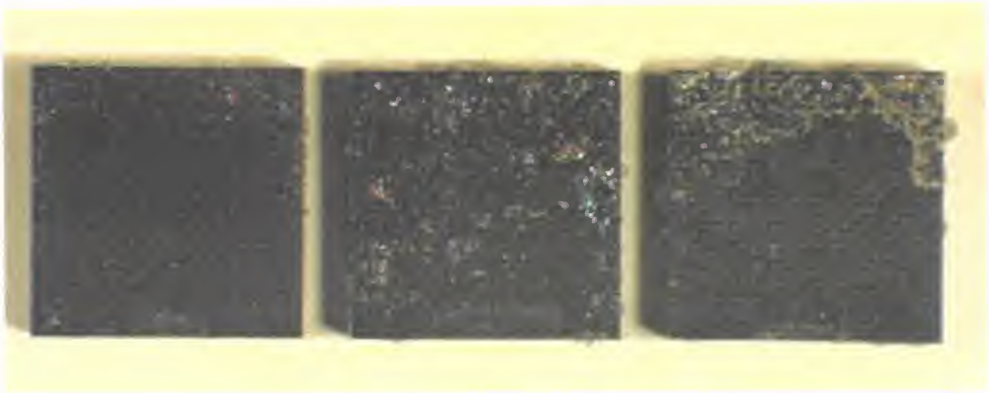
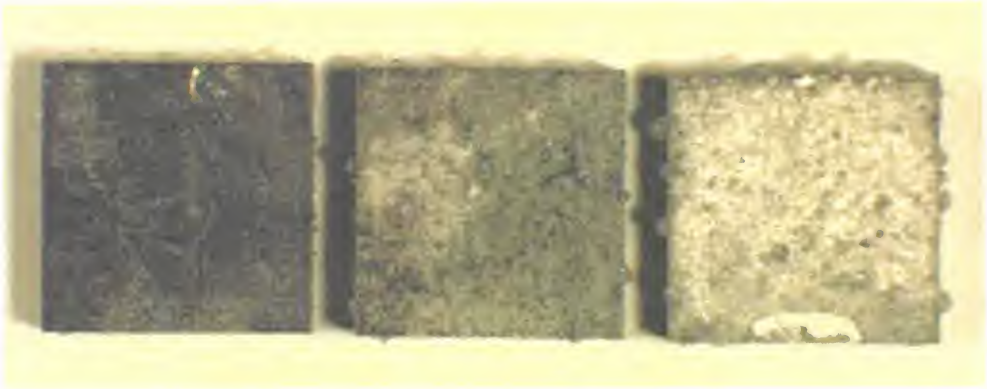
UHTCC After 1627°C Furnace Oxidation

C/SiC Side



HfB_2 Side



Cycles	1	5	10
Hours	0.167	0.833	1.667

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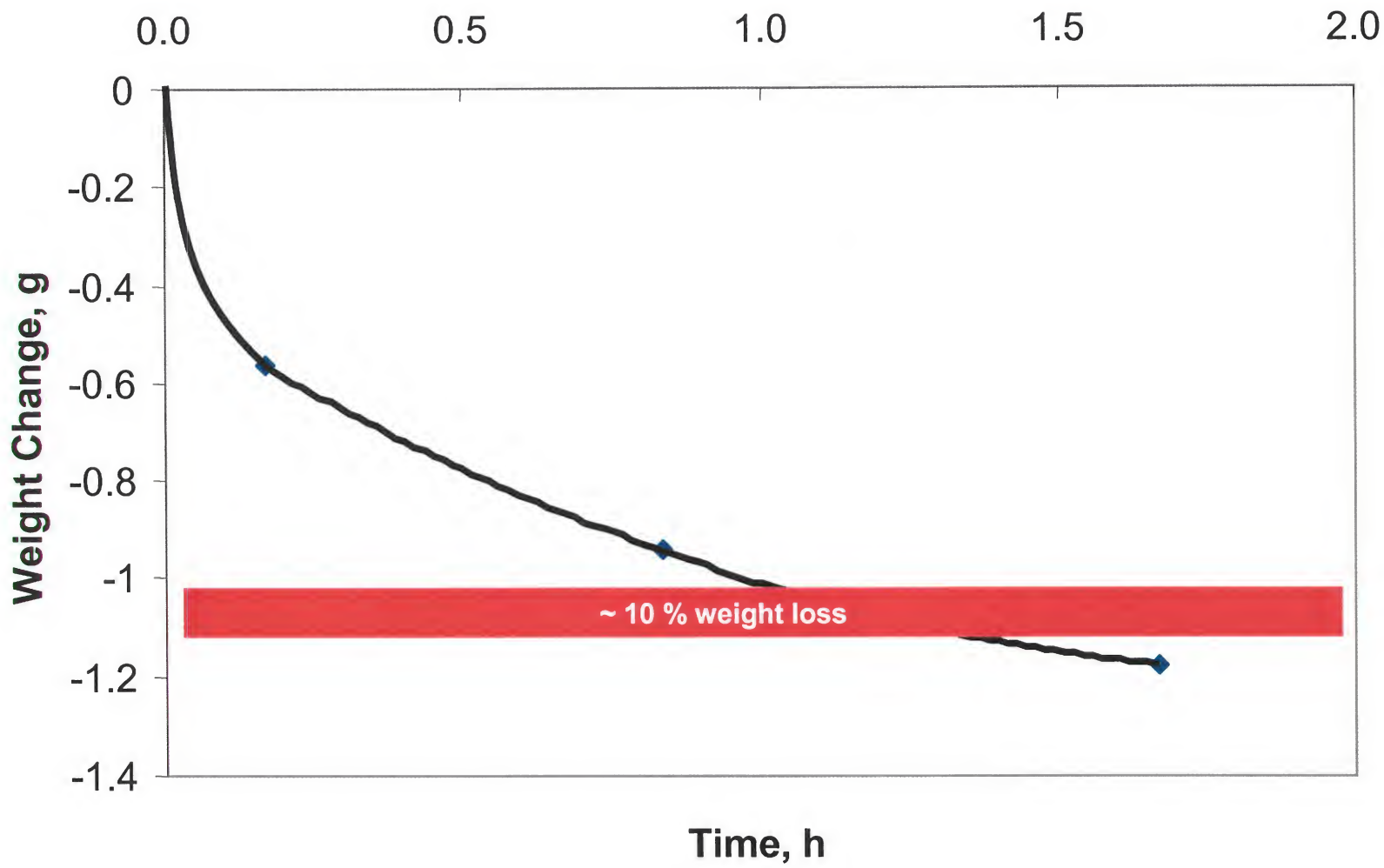
Furnace Oxidation of UHTCC at 1627°C



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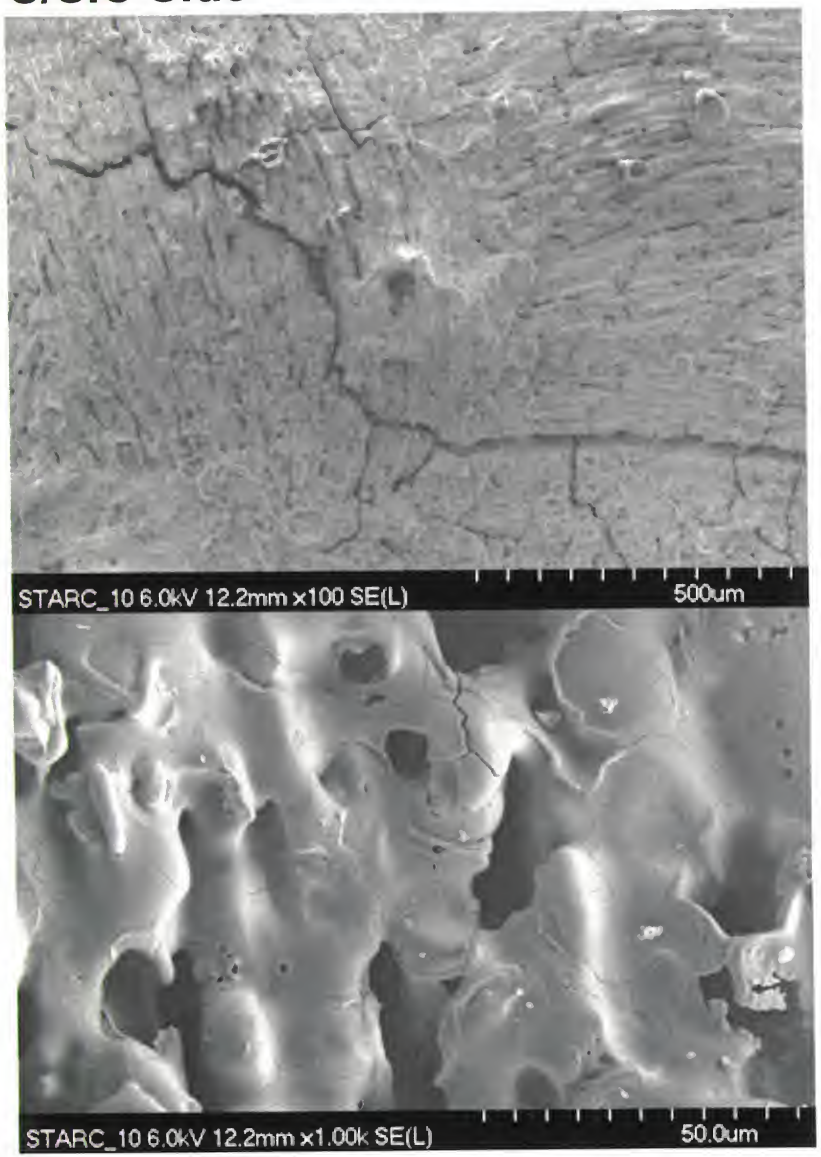
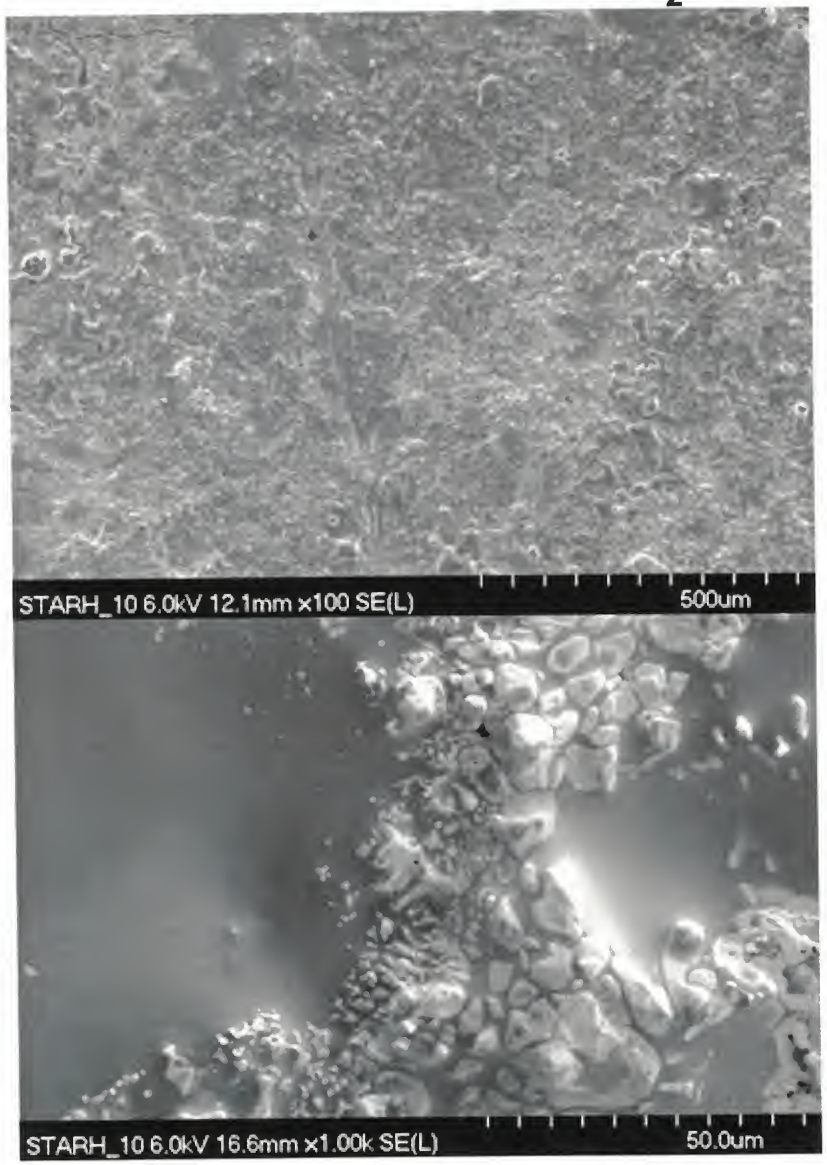
UHTCC After 10 Ten-minute Cycles in Air at 1627°C

C/SiC Side



HfB_2 Side



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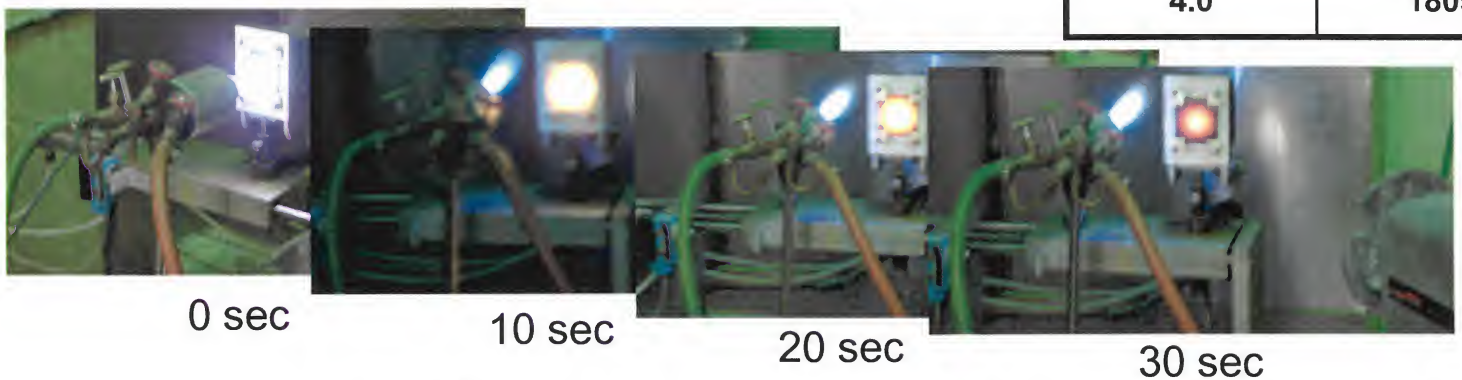
UHTCC Oxy-Acetylene Torch Test

- **Sample O: One 4 min. cycle to 1805°C**
  - Temps with Ircon 2 color pyrometer, 980-1760°C range
  - Weight loss = 0.79 g, or 1.9%
- **Sample N: Three ~4 min. cycles**
  - Cycle 1 max temp 1815°C
  - Cycle 2 max temp 1915°C
  - Cycle 3 max temp 2015°C
  - Weight loss = 1.95 g, or 4.6%

Sample O	
Time, min.	Temp, °C
0.5	1720
1.0	1750
1.5	1750
2.0	1755
2.5	1765
3.0	1775
3.5	1790
4.0	1805



Specimen "O" during cool-down

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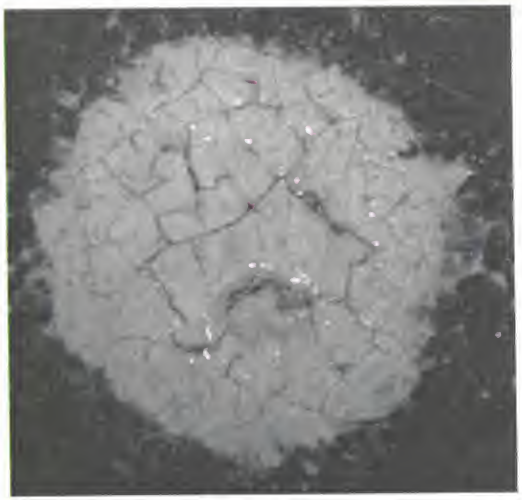
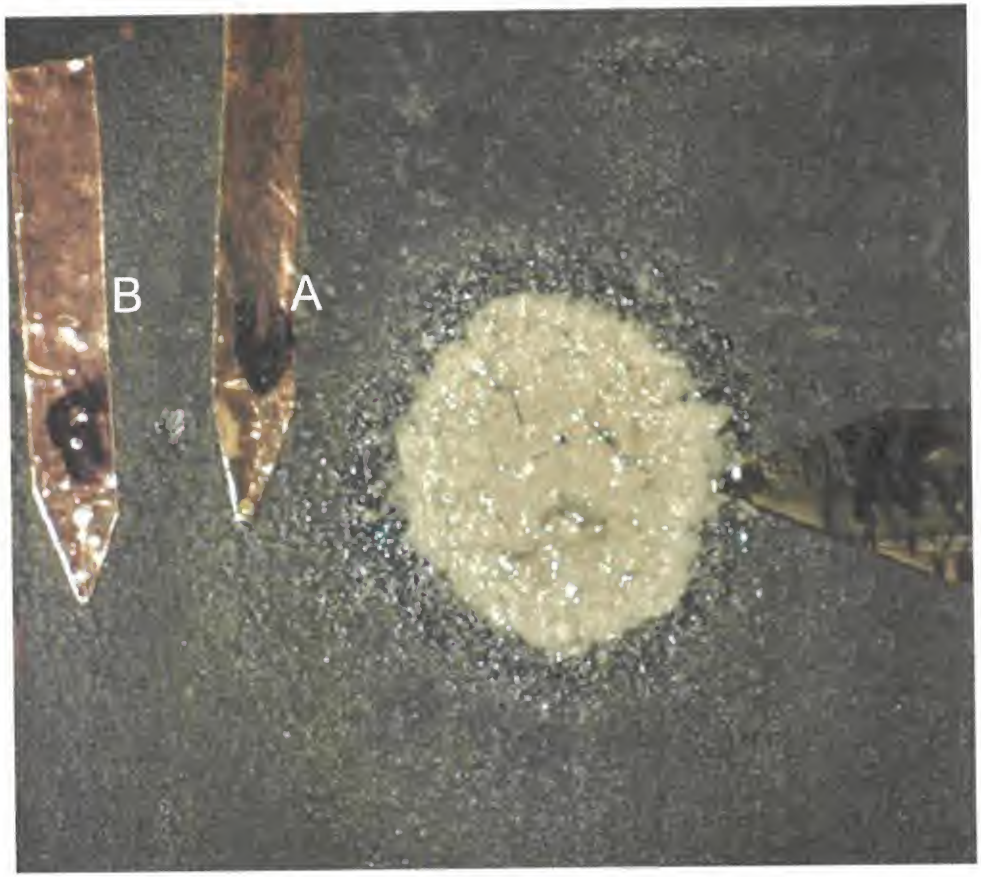
UHTC Surface After Three 4-Minute Torch Cycles to 1815 to 2015°C





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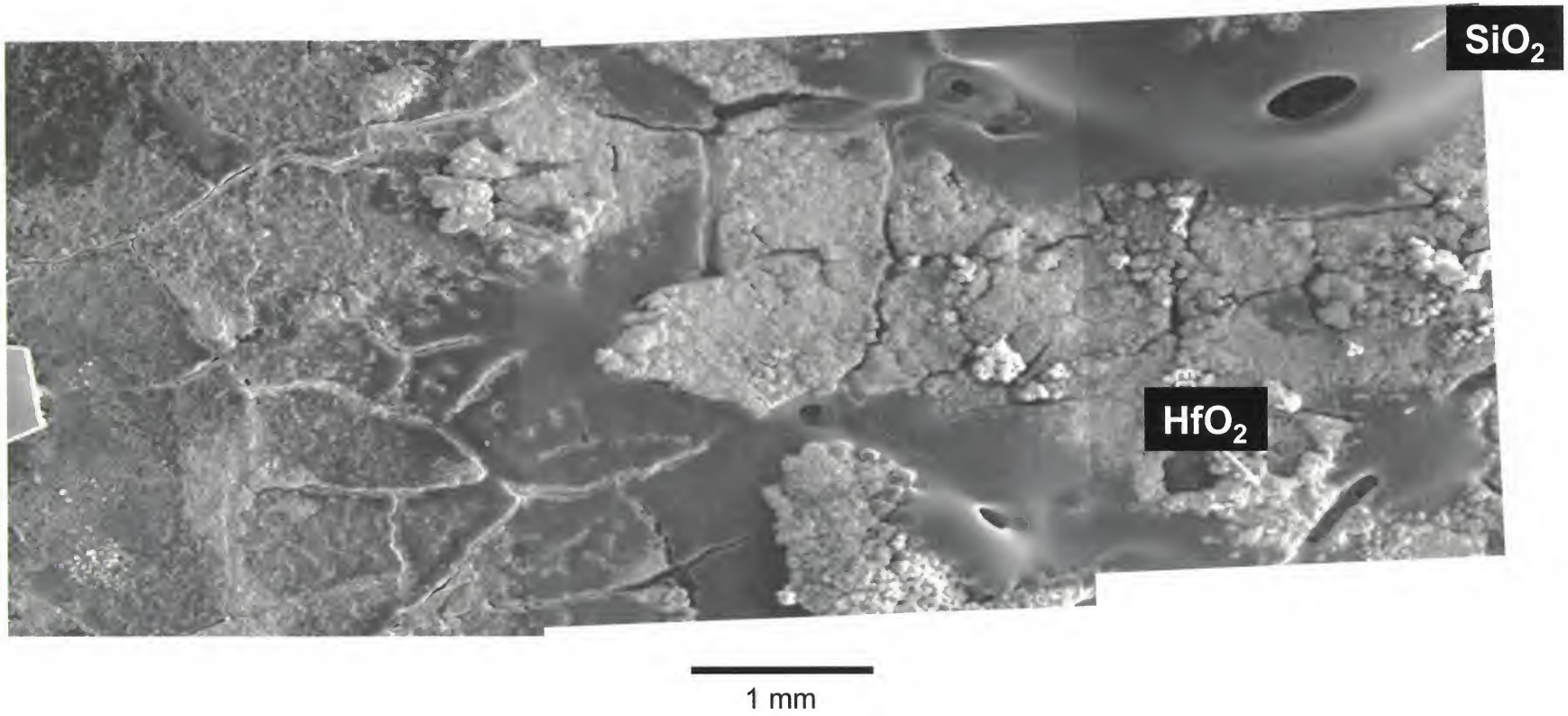
SEM of Center of UHTC Surface After Three Torch Cycles



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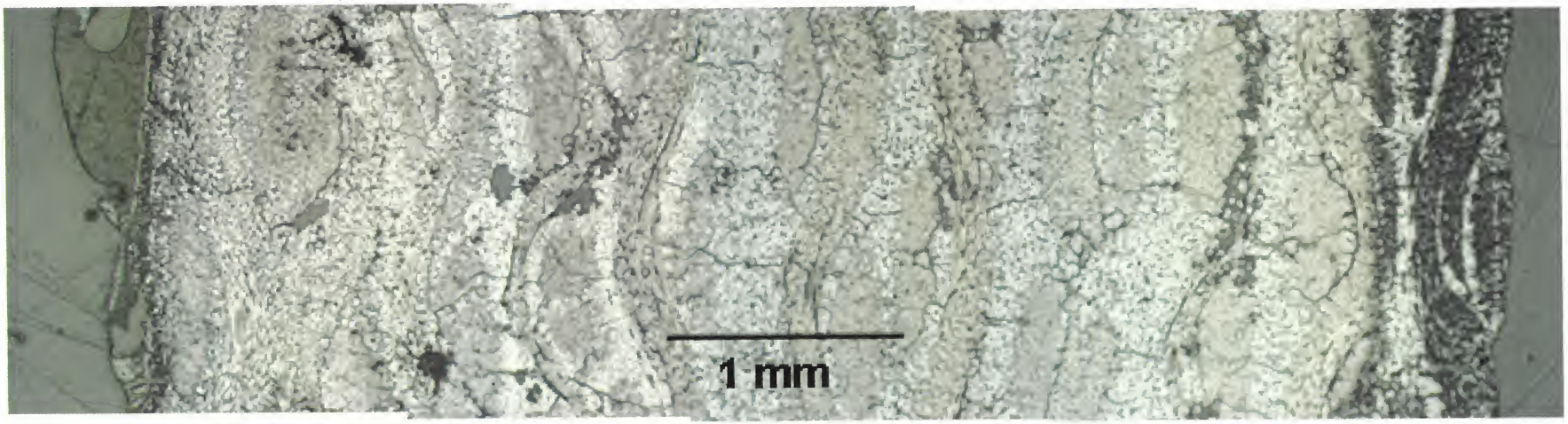
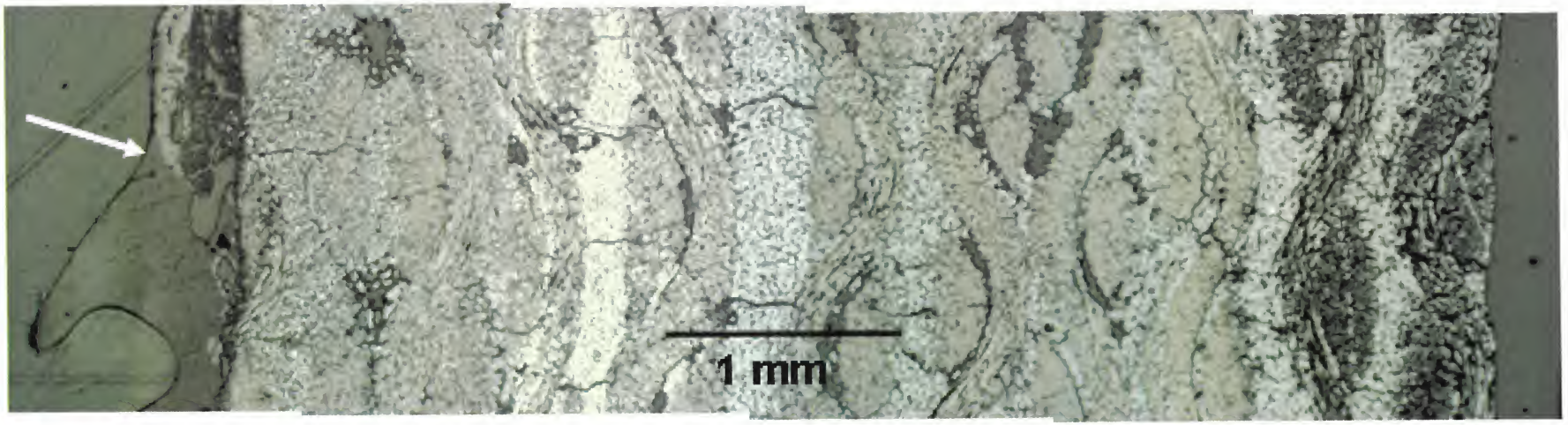
UHTCC Specimen Center Cross Sections After Torch Test

HfB_2 C/SiC



One 4-Minute Cycle to 1805°C



Three 4-Minute Torch Cycles to 1815 to 2015°C

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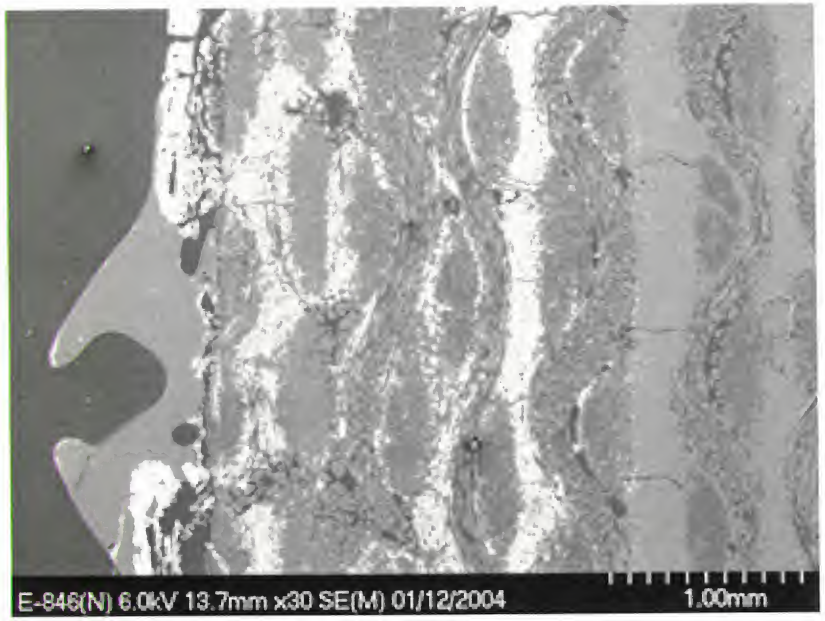
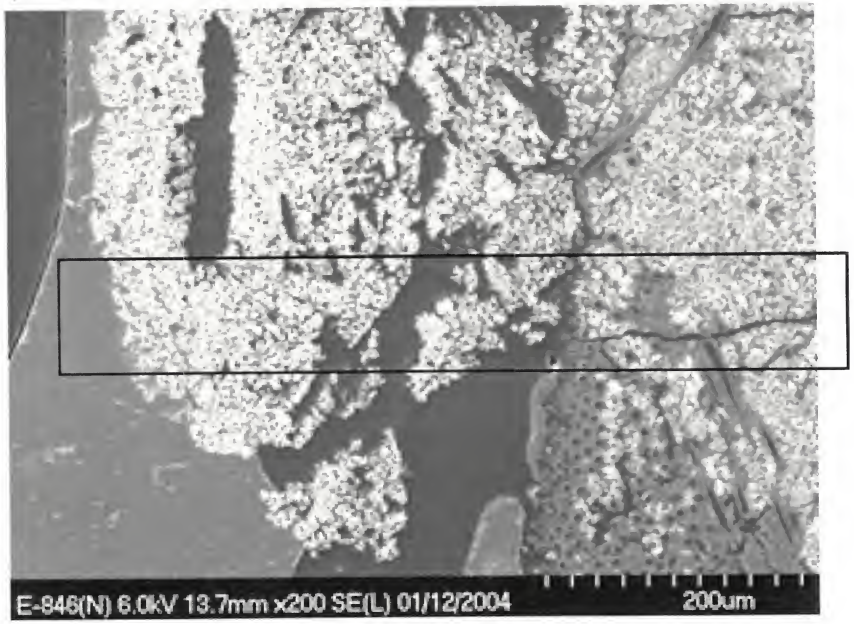
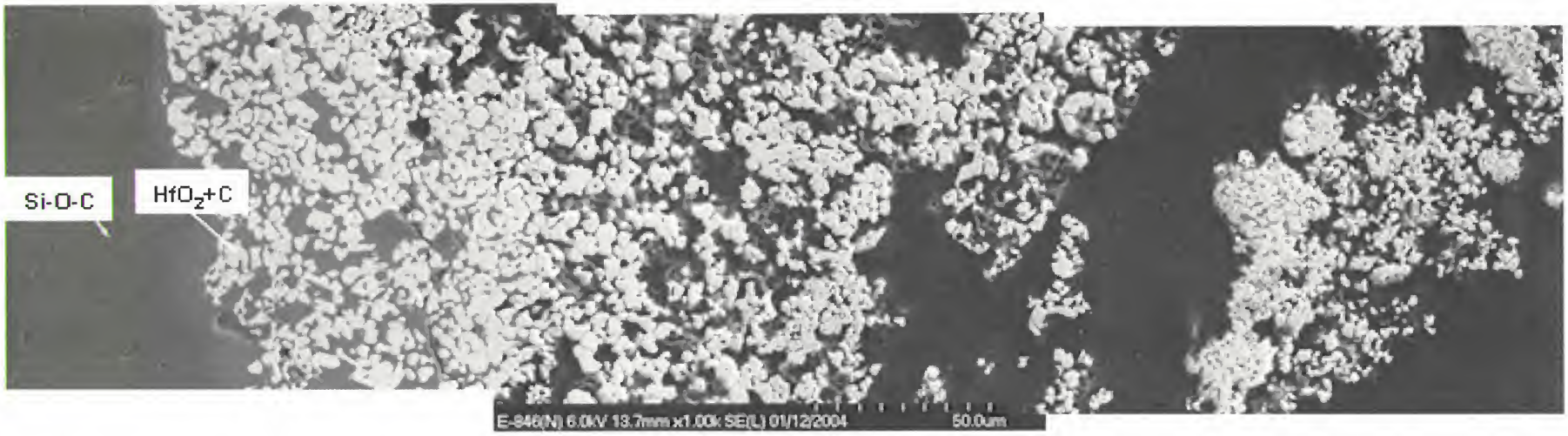
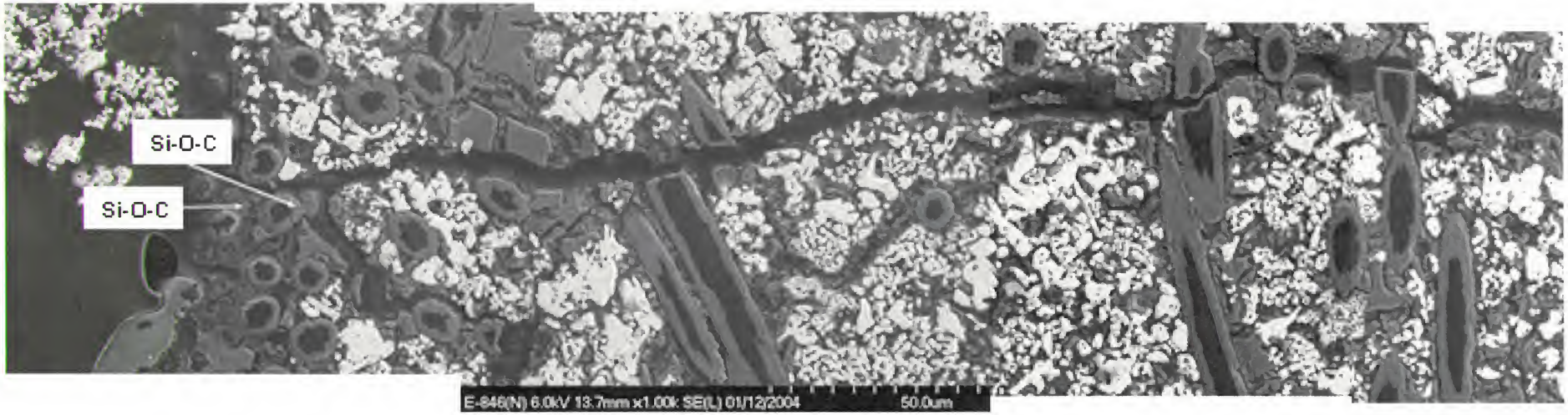
UHTC Side, Center Hot Zone





UHTC Side, Center Hot Zone





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Conclusions

- **$TaSi_2$ Additions**
  - **Good results at 1627°C, but oxide scale too fluid at > ~ 1800°C**
  - **Look at lower levels of additive**
- **UHTCC**
  - **Processing**
    - Uniform and through thickness graded microstructure achieved
    - Matrix cracking due to thermal expansion mismatch between C fibers and matrix constituents is a concern
  - **Mechanical Properties**
    - Flexural strength was close to expected values based on rule of mixtures with no matrix contribution
    - Some evidence of composite behavior
  - **Furnace Oxidation**
    - Weight loss and metallography indicated that carbon fiber oxidation occurred rapidly
  - **Torch Test**
    - Based on observed temperature spikes during test, adherence of the HfO_2-rich scale is an area of concern
    - Material withstood ~2000°C (~3600°F), severe heat-up and thermal gradients with no major visible distress, but with significant HfB_2 coating and carbon fiber oxidation revealed by metallography

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Future Work

- **Complete oxidation studies focused on lower levels of Ta addition to achieve doping without low melting phase formation**

- **Continue UHTCC evaluation**
  - Complete metallography on Starfire specimens
  - Evaluate other NASA and industry developed materials

- **Continue UHTCC development**
  - Fiber coatings need to be incorporated to address fiber oxidation issues
  - Advanced SiC fibers need to be evaluated to address oxidation and thermal expansion mismatch issues

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